US ERA ARCHIVE DOCUMENT

## **NALED**

# **Product Chemistry**

# Task 1: Review and Evaluation of Individual Studies

Contract No. 68-01-5830

Final Report

October 29, 1982

#### Submitted to:

Environmental Protection Agency Arlington, Virginia 22202

#### Submitted by:

Dynamac Corporation Enviro Control Division The Dynamac Building 11140 Rockville Pike Rockville, MD 20852

## NALED

## PRODUCT CHEMISTRY

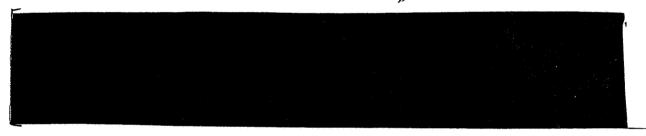
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Study	
1	Composition of technical naled.
2	Manufacturing process for technical naled.
3	Analytical methods for the determination of naled in the technical product.
4	Names, identity, and physical/chemical properties of technical naled.

CASE GS 0092	NALED	STUDY 1	12/22/8	T PM 110
CHEM 034401	Na	led		
BRANCH RCBR	DISC 05 TOPI	C 05 GUIDEL	INE 40 CFR 163.61-8	
FORMULATION 00	- ACTIVE INGREDIEN	Т		
FICHE/MASTER ID	00074653	CONTENT CAT 12		*****
Chevron Chemical Chemical: Di	Company (1966) Na brom. (Unpublish	ed study received S	ty and Composition of ep. 12, 1966 under 7F0	532; CDL:092821-H
FICHE/MASTER ID	00065493	CONTENT CAT 12		
Mar. 19, 1976	under 239-2444; C	DL:229289-F).	cal Naled. (Unpublish	
		CONTENT CAT 06		
Dibrom. (Unp		eived Mar. 19, 1976 CDL:229289-H).	Determination of Impu under 239-2444; submi	tted by Chevron
SUBST. CLASS =				and in the pass are the first open and the pass and the pass are the pass and the pass are the pass and the pass are the p
DIRECT RVW TIME		START-DATE May 7	END DATE	May 10
ORG: D	taff Scientist	ro Control Division	, Rockville, MD	
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APPROVED BY: TITLE: ORG: LOC/TEL:				
SIGNATURE:			DATE:	

Three confidential business documents submitted by Chevron Chemical Co. provide information on the composition of technical naled. The available data partially satisfy data requirements in Sections 163.61-5 and 163.61-6 of EPA's Proposed Guidelines for Registering Pesticides (July 1978) by providing the identity of potential impurities. However, data gaps include: i) unacceptable or inadequately described methods; ii) sensitivities and detection limits were not reported for any of the methods used; iii) no indication was made as to whether samples of technical naled were from different batches; iv) early and later documents report two different naled limits; v) several impurities that could be present at  $\ge 0.1\%$  (by weight of technical) were not identified or quantified individually; and vi) certification of limits was not made.

#### Methods and Materials:



#### Reported Results:

# QUALITY CONTROL PROCEDURE INFORMATION IS NOT INCLUDED

An early document of Chevron Chemical Co. (1966, 00074653) showed that 10 samples of technical naled (source: Shell Chemical Co.) contained the active ingredient at Later documents (Chevron Chemical Co., 19??, 00065493 and 1971, 00065494) reported the limits as The impurities in technical naled were reported.

#### Discussion:

The available data only partially satisfy data requirements (identity of potential impurities) because: i) all methods were either unsuitable or inadequately described; ii) sensitivities and detection limits were not reported for any of the methods used; iii) there was no indication that samples of technical naled were from different batches; iv) there is a discrepancy between early and later reports regarding the naled limits; and v) several impurities that could be present at  $\ge 0.1\%$  (by weight of technical) were not identified or quantified individually.

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CASE GS 0092	NALED	STUDY	2	12/22,	/81 	PM	1 110
CHEM 034401	Na	led					
BRANCH RCBR	DISC 05 TOPIC	05	GUII	DELINE 40 CFR	163.6	1-8	
FORMULATION (	OO - ACTIVE INGREDIEN	T					
FICHE/MASTER	ID 00074653	CONTENT	AT 12				
Chevron Chemical: 092821-H).	cal Company (1966) Na Dibrom. (Unpublish	me, Chemical ed study rec	Identity eived Sep	and Composit: . 12, 1966 und	ion of der 7F	the 1	Pesticide CDL:
FICHE/MASTER	ID 00074791	CONTENT	CAT 06				
Chemical Chem Manufactur 232095).	ical Company (19??) I	lished study	received	-dichloroethy	l Dime under	thyl 239-	Phosphate): 1633; CDL:
SUBST. CLASS							
DIRECT RVW T	IME = 1 (MH)	START-DATE	May 7		DATE	May	7
ORG:	N. Hajjar Staff Scientist Dynamac Corp., Envir 468-2500		vision, R	ockville, MD	DATE:	May	7, 1982
APPROVED BY:							
TITLE: ORG: LOC/TEL:							
SIGNATURE:		a.			DATE:		

Two confidential business documents submitted by Chevron Chemical Co. (00074653 and 00074791) describe the process used to manufacture technical naled. Purities of starting materials are provided. The available information satisfies data requirements in Section 163.161-4 of EPA's Proposed Guidelines for Registering Pesticides (July 1978).

#### Methods and Materials:

The procedures used to manufacture technical naled, including the purity of certain starting materials, are presented.

#### Reported Results:

N/A.

#### Discussion:

N/A.

CASE GS 0092 NALED STUDY 3 PM 110 12/22/81 CHEM 034401 Naled DISC 05 TOPIC 1015 GUIDELINE 40 CFR 163.61-7 BRANCH RCBR FORMULATION 01 - TECHNICAL CHEMICAL \_\_\_\_\_ CONTENT CAT --FICHE/MASTER ID GS092006 Carlstrom, A.A. 1975. Gas-liquid chromatographic determination of naled in pesticide formulations. JAOAC 58(6):1162-1168. FICHE/MASTER ID 00074846 CONTENT CAT --Chevron Chemical Company. 1964. Analysis of Dibrom. Method [D-IV] dated July 21. 1964. Unpublished study received June 23, 1965 under unknown admin. no.; CDL:102845-A. CONTENT CAT 06 FICHE/MASTER ID 00074655 Chevron Chemical Company. 1966. Ortho method of analysis--D-IX-a: Dibrom (R) Naled by gas chromatograph. Method dated May 27, 1966. Unpublished study received Sep. 12, 1966 under 7F0532; CDL:092821-J. CONTENT CAT 06 FICHE/MASTER ID 00074724 Ospension, N.J. 1958. Letter sent to G.K. Kohn dated Feb. 4, 1958: Dibrom-physical and chemical properties. Includes method dated Apr. 3, 1957. Unpublished study received Feb. 10, 1958 under unknown admin. no.; submitted by Chevron Chemical Co., Richmond, Calif.; CDL: 119717-A. SUBST. CLASS = S. DIRECT RVW TIME = 4 (MH) START-DATE May 10 END DATE May 10 REVIEWED BY: N. Hajjar TITLE: Staff Scientist Dynamac Corp., Enviro Control Division, Rockville, MD 468-2500 / LOC/TEL: huiles & Hojan DATE: May 10, 1982 SIGNATURE: APPROVED BY: TITLE: ORG: LOC/TEL: DATE: SIGNATURE:

Three Chevron Chemical Co. documents and one published study describe analytical methods for the determination of naled in technical and formulated products. A method involving the potentiometric titration of Bris inadequate because it is not specific for naled (00074724). The GC methods (00074655, 00074846, and GS092006) also do not satisfy data requirements in Section 163.61-7 of EPA's Proposed Guidelines for Registering Pesticides (July 1978) for the following reasons: detection limits and validation data were not provided, data for technical naled or formulated products were not reported, and a confirmatory procedure was not detailed. An update of the method(s) used for quality control of the technical and formulated products of naled is required.

#### Methods and Materials:

Chevron Chemical Co. (Ospenson, 1958, 00074724)

GC Method D-IX-a (Chevron Chemical Co., 1966, 00074655)

GC Method D-IV (Chevron Chemical Co., 1964, 00074846)

AOAC GC method, 1975, GS092006 QUALITY CONTROL PROCEDURE INFORMATION IS NOT INCLUDED

Chloroform-soluble material in samples of technical naled and formulated products is analyzed by using a GC equipped with a hydrogen flame detector. A column containing 3% XE-60 on Gas-Chrom Q is operated at 135°C. Pretreatment by repeated injections of the naled-dibutyl phthalate standard is required. Several pesticides were tested to determine if they interfere with naled determinations.

#### Reported Results:

The recovery of the Br titration method was 80%; the detection limit was not reported. There was also an indication that homogeneous solutions of naled were not obtained.

Detection limits, validation data, and analytical data were not provided for any of the GC methods. It was suggested that typical peak shapes may not be reliably obtained by using Method D-IX-a. The type of detector used for Method D-IV was not reported. Benzene hexachloride, diazinon, and dicrotophos interfere with either naled or dibutyl phthalate using the AOAC method.

#### Discussion:

N/A.

CASE GS 0092 NALED STUDY 4 PM 110 12/22/87 UNDER GO OUDE HALLD DIED TO THE TOTAL THE TOTA CHEM 034401 Naled BRANCH RCBR DISC 05 TOPIC 05 GUIDELINE 40 CFR 163.61-8 FORMULATION 01 - TECHNICAL CHEMICAL FICHE/MASTER ID 00074790 CONTENT CAT 12 Chevron Chemical Company. 1965? Product chemistry data for Chevron naled technical. Unpublished study received Oct. 17, 1977 under 239-1633; CDL:232095-A. CONTENT CAT 12 FICHE/MASTER ID 00074653 Chevron Chemical Company. 1966. Name, chemical identity, and composition of the pesticide chemical: Dibrom. Unpublished study received Sep. 12, 1966 under 7F0532; CDL:092821-H. CONTENT CAT --FICHE/MASTER ID GS092040 Chevron Chemical Company. 1974. Chevron-Naled: Formulator's manual. Unpublished study received Oct. 22, 1974 under 239-1633, Accession No. 233083. CONTENT CAT 06 FICHE/MASTER ID 00074724 Ospenson, J.N. 1958. Letter sent to G.K. Kohn dated Feb. 4, 1958: Dibrom--physical and chemical properties. Includes method dated Apr. 3, 1957. Unpublished study received Feb. 10, 1958 under unknown admin. no.; submitted by Chevron Chemical Co., Richmond, Calif.; SUBST. CLASS = S. DIRECT RVW TIME = 10 (MH) START-DATE May 6 END DATE May 7 REVIEWED BY: N. Hajjar TITLE: Staff Scientist ORG: Dynamac Corp., Enviro Control Division, Rockville, MD LOC/TEL: 468-2500 / miles & Hoyan DATE: May 7, 1982 SIGNATURE: APPROVED BY: TITLE: ORG: LOC/TEL: DATE: SIGNATURE:

Four documents submitted by Chevron Chemical Co. (00074635, 00074724, 00074790, and GS092040) provide the names, identity, and physical/chemical properties of technical and pure naled. The information pertaining to technical naled partially satisfies data requirements in Sections 163.61-3 and 163.61-8 of EPA's Proposed Guidelines for Registering Pesticides (July 1978).

#### Methods and Materials:

N/A.

#### Reported Results:

The names, identity, and physical/chemical properties of technical naled, and in some cases pure naled, are presented below.

Name, identity, and physical/chemical properties of naled.

Chemical name: a,b,c	1,2-Dibromo-2,2-dichloroethyl dimethyl phosphate.
Common name: a,b,c	Naled.
Trade name: a,b,c	Dibrom.
U.S. Patent Number: b,c	2,971,882
Empirical formula: a,b,c	C4H7O4PBr2Cl2
Molecular weight: b,c,d	381
a,b,c,d	H <sub>3</sub> O    H Br
	$\rightarrow$ P-0-C-C-C1
C	H <sub>3</sub> 0 Br C1
Molecular composition	Theoretical Found
(% by weight):	н 1.84 1.98
( by weight).	P 8.13 8.30
	C 12.62 12.63
•	C1 18.62 18.30
	Br 40.0 40.9
Total hal	ogen 10.5 meq/g 10.49 meq/g
Melting point (°C):	$25.56^{\text{C}}$ (27-28.5 <sup>a</sup> , <sup>b</sup> , 25.5-26.5 <sup>d</sup> for pure compound).
n :7:	$110^{\circ}$ on $120^{\circ}$ , b at 0.5 mmHg (108-110 at 0.25 mmHg for

Boiling point (°C):

 $110^{\text{C}}$  or  $120^{\text{a}}$ , b at 0.5 mmHg (108-110 at 0.25 mmHg for pure compound d).

Density/specific gravity (temperature not provided unless specified):

1.97 at  $20^{\circ}C^{\circ}$ ; 1.96<sup>a,b</sup>; 1.465<sup>a</sup> (1.96 for pure compound<sup>d</sup>). 2 x  $10^{-4}$  mmHg at  $20^{\circ}C$ .

Vapor pressure: a,b,c

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Refractive index:

 $n_{\rm n}^{28} = 1.5108^{\rm b,d} \text{ or } \sim 1.512^{\rm c}$ .

Infrared absorption:

Not reported for technical (the pure compound absorbed strongly at  $\sim 860$ , 900, 1020, 1060, 1120, 1280, and 1295 cm<sup>-1</sup> as determined from the

presented spectrum).

Viscosity:

22 cp at 20°C<sup>a</sup> or 210-250 sec Saybolt at 37.8°C<sup>C</sup>

Physical state: a,c

Oily liquid (pure compound is a low melting point solid).

Color: a,c

Light, straw-colored (white when pure).

Odor: a,c

Slightly pungent.

Miscibility:

Miscible in most organic solvents; emulsifiable in water.

Solubility: a,b,c

Limited solubility (1-5%) in aliphatic solvents; highly soluble in oxygenated solvents such as ketones and alcohols; low solubility in water  $(0.2 \text{ g/}100 \text{ ml at } 23.3^{\circ}\text{C}^{a})$ .

Stability:

Saturated aqueous solutions of technical naled (2 g/1) are rapidly hydrolyzed at 23.3°Ca (naled of unspecified purity in aqueous solution is not hydrolyzed at 0°C and has half-lives of  $\sim 13$  and  $\sim 2$ days at 25 and 40°C, respectivelyb,d complete hydrolysis occurs under reflux within 4 hours in distilled water or in the presence of excess NaOHO).

Flashpoint: a

53°C.

Dissociation constants:

N/A (according to Chevron Chemical Co.).

pH: a

N/A (according to Chevron Chemical Co.).

Explosiveness:

Not explosive.

Corrosiveness:

Corrosive to steel, aluminum, and magnesiuma; corrosive to all the above as well as copper and brass when in the presence of water<sup>C</sup>.

Oxidizing/reducing action: a Storage stability:

Not an oxidizing or reducing agent.

No breakdown of the technical material (source: Shell Chemical Co.) occurred within 18 months at ambient temperatures and 2.5% degradation occurred at 38°C over the same period.

<sup>&</sup>lt;sup>a</sup>From 00074790.

DFrom 00074653.

CFrom GS092040.

dFrom 00074724.

Discussion:

N/A.

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